EXHIBIT J

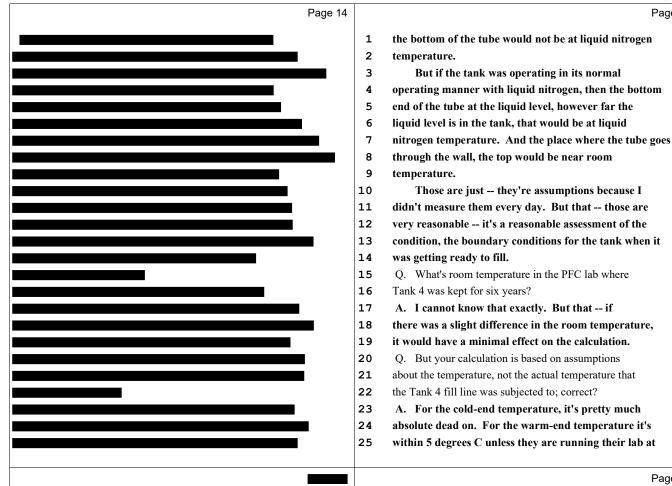
CE	MIER LITIGATION	,	December 14, 2020
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1	UNITED STATES DISTRICT COURT	1	INDEX
2	NORTHERN DISTRICT OF CALIFORNIA	2	
3	SAN FRANCISCO DIVISION	3	EXAMINATION BY: PAGE
4		4	MS. ZEMAN 6
5		5	00
6		6	
7	IN RE PACIFIC FERTILITY) CENTER LITIGATION,) Case No. 3:18-cv-01586-JSC	7	EXHIBITS
8	}	8	PLAINTIFFS' EXHIBIT NO. DESCRIPTION PAGE
9		9	440 December 4, 2020, Rebuttal Report Of Franklin K. Miller, Ph.D.,
10		10	Franklin K. Miller, Ph.D., (12 Pages) 6
11		11	441 December 11, 2020, Rebuttal Report Of
12		12	Franklin K. Miller, Ph.D., (19 Pages) 70
13	VIDEOTAPED & VIDEOCONFERENCED DEPOSITION of	13	442 "Nitrogen Adsorption Isotherms For
14	FRANKLIN K. MILLER, Ph.D., taken on behalf of Plaintiffs	14	Zeolite And Activated Carbon" Paper By L.C. Yang, T.D. Vo, And
15	remotely beginning at 9:32 a.m., Monday, December 14,	15	H.H. Burris, (10 Pages) 75
16	2020, before CHERREE P. PETERSON, RPR, CRR, Certified	16	00
17	Shorthand Reporter No. 11108.	17	
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25		25	
	Page 3		Page 5
1	APPEARANCES		3
2		1	DECEMBER 14, 2020
3	FOR THE PLAINTIFFS:	2	oOo
4	GIBBS LAW GROUP LLP	3	BE IT REMEMBERED that set on Monday, the 14th
5	505 14th Street, Suite 1110 Oakland, California 94162	4	day of December, 2020, commencing at the hour of 9:32
6	BY: AMY M. ZEMAN, ESQ. (510) 350-9700	5	a.m., taken remotely before me, Cherree P. Peterson,
7	amz@classlawgroup.com	6	RPR, CRR, CSR No. 11108, a Certified Shorthand Reporter,
8	FOR THE DEFENDANT CHART INDUSTRIES, INC.:	7	personally appeared
9	SWANSON, MARTIN & BELL, LLP	8	FRANKLIN K. MILLER, Ph.D.,
10	SWANSON, MARTIN & BELL, LLP 330 N Wabash, Suite 3300 Chicago, Illinois 60611	9	having been called as a witness by the plaintiffs, who
11	BY: JOHN J. DUFFY, ESQ. KEVIN M. RINGEL, ESQ.	10	having been duly sworn by me to tell the truth, the
12	(312) 321-9100	11	whole truth, and nothing but the truth, was thereupon
13	jduffy@smbtrials.com kringel@smbtrials.com	12	examined and testified as hereinafter set forth.
	MUE VIDEOCDA DUED.	13	00
14	THE VIDEOGRAPHER:	14	THE VIDEOGRAPHER: Good morning, Counsel. My
15	PHILIP KNOWLES	15	name is Philip Knowles. I am the host and videographer
16		16	associated with Barkley Court Reporters located at 201
17		17	California Street, Suite 375, in San Francisco,
18		18	California 94111. The date today is Monday, December
19		19	14th, 2020, and the time is approximately 9:33 a.m.
20		20	Pacific Standard Time.
21		21	This deposition is taking place remotely via
22		22	Zoom in the matter of Pacific Fertility Center
23		23	litigation with case number 3:18-CV-01586-JSC. This is
			4 '1 / 11 '/' OD D 11' Y Y Y Y Y
24		24	the videotaped deposition of Dr. Franklin K. Miller
		24 25	the videotaped deposition of Dr. Franklin K. Miller being taken on behalf of plaintiffs.

CE	NIEKLINGATION		December 14, 2020
	Page	6	Page 8
1	Will counsels for the parties please voice	1	A. Yes, it does.
2	identify themselves.	2	Q. Is that a basic concept a trained engineer
3	MS. ZEMAN: This is Amy Zeman on behalf of the	3	would understand?
4	plaintiffs.	4	A. Yes, it is.
5	MR. DUFFY: John Duffy on behalf of Chart.	5	Q. Do you need to do any testing to know that
6	MR. RINGEL: Kevin Ringel also for Chart.	6	metal contracts when it is cooled?
7	THE VIDEOGRAPHER: Thank you. The court	7	A. You well, it's a bit complicated. You would
8	reporter may now answer in the witness and make a	8	someone has to do testing to know how much metal
9	statement for the record.	9	contracts. There are tables of values for equations
10	THE REPORTER: Raise your right hand, please,	10	that are used to describe the quantity or the amount
11	Doctor.	11	that metal contracts when it changes temperature. So
12	(Whereupon the witness was placed under oath.)	12	Q. You mean go ahead.
13	THE REPORTER: Thank you.	13	A. So if you want to know that metal contracts,
14	EXAMINATION BY MS. ZEMAN	14	no, you don't have to know do a do a test. But
15	Q. Good morning, Dr. Miller. Good to see you	15	there are some materials that contract and expand as
16	again. Thank you for joining us.	16	they go down in temperature, as I have direct experience
17	A. Good morning.	17	with that. Some materials get shorter and then when
18	Q. I will try not to take up too much time of your	18	they get to a certain temperature they start getting
19	day today, but I do have a few questions about the two	19	longer again. But that's not specifically metal.
20	reports you've put in since we spoke last.	20	So you could know it would contract, but you
21	For starters, I would like the Exhibit 440 to	21	couldn't quantify how much unless you have test data
22	be entered and made available to you. That will be your	22	from someone or from a handbook from tables. And that
23	rebuttal report dated December 4th of 2020.	23	contraction is a function of the temperature.
24	(Plaintiffs' Exhibit 440 marked for	24	Metal does not contract the same amount at room
25	identification.)	25	temperature per degree of temperature change as it
	,		
	Page	7	Page 9
1	-		
1 2	MS. ZEMAN: Philip, are you going to?	1	contracts at low temperature per degree per temperature
2	MS. ZEMAN: Philip, are you going to? THE VIDEOGRAPHER: Yeah. Amy, what I did with	1 2	contracts at low temperature per degree per temperature change. So that's a temperature-dependent material
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2 3 4	MS. ZEMAN: Philip, are you going to? THE VIDEOGRAPHER: Yeah. Amy, what I did with Dena last week was I made it, and then I saved it, and then I sent it back into the chat, and then I screen	1 2 3 4	contracts at low temperature per degree per temperature change. So that's a temperature-dependent material property. Q. Dr. Miller, does metal expand when it is
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2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22	MS. ZEMAN: Philip, are you going to? THE VIDEOGRAPHER: Yeah. Amy, what I did with Dena last week was I made it, and then I saved it, and then I sent it back into the chat, and then I screen shared. Would you like me to do the same thing? MS. ZEMAN: All but the screen share. THE VIDEOGRAPHER: Okay. And you know what, that makes sense. I didn't screen share for Dena. So sorry, guys. It just takes a second to save it. And sorry for all the unnecessary commentary, Cherree. Okay. Exhibit 440 is available for download. Q. BY MS. ZEMAN: Dr. Miller, if you could download that and take a look at it and let me know if you recognize the document. A. Yes, I recognize the document. Q. Okay. And what is this? A. This is my December 4th rebuttal report. Q. Okay. Great. And I don't think I have a question on it directly right away. But if at any point during my questions if you want to refer to that, you are welcome to do so. And there will be times when I do refer to specific references in that report.	1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22	contracts at low temperature per degree per temperature change. So that's a temperature-dependent material property. Q. Dr. Miller, does metal expand when it is warmed? A. Yes, it does. Q. Is that a basic concept a trained engineer would understand? A. Yes, it is. Q. If the metal at a joint expands or contracts do basic engineering principles dictate that the joint will experience stress? A. Actually, no, the joint will not experience stress if the temperature changes at a joint. If the entire joint changes temperature uniformly at the same time, there will be a change in the dimension of the material but no stress as a result of that change. Q. And that example that you just described, that would be if all of the parts at the joint are free to expand; correct? A. It has nothing to do with free to expand. If I take a piece of metal and I uniformly change its

	Page 10	
1	of that and another, there will be no stress induced,	
2	thermally induced stress in that part.	
3	Q. Does that apply if there are different types of	
4	metal involved in the joint?	
5	A. If there are different types of metal involved	
6	in some buildup when you say by joint, if something	
7	is welded then it's relatively uniform in composition.	
8	There are rare cases where we do weld copper to	
9	stainless steel, those kind of joints. Those are very	
10	rare though.	
11	But generally speaking if you have a welded	
12	part, that joint is uniform relatively uniform	
13	composition. In we often do calculations in	
14	cryogenics where we have stack ups of parts when we go	
15	cold. For instance, we may have copper or brass flanges	
16	bolted together by stainless steel screws. And when we	
17	go cold, we like to assure that they're getting tighter,	
18	not looser. And in that case one material contracts at	
19	a different rate than another one. And often what we'll	
20	do is use brass screws with copper flanges because their	
21	coefficients of thermal expansion are very similar	
22	versus something like stainless steel.	
23	And then there are ways to compensate that we	
24	- and part of cryogenic design is to use Invar washers,	
25	Bellville washers that are spring washers to take up	
	Page 11	
	1 495 1 1	
1	that slack.	
2	So yes, you can have changes in dimensions with	
3	the same temperature, but that's because the composition	
4	of the material is significantly different and has a	
5	different material property behavior.	
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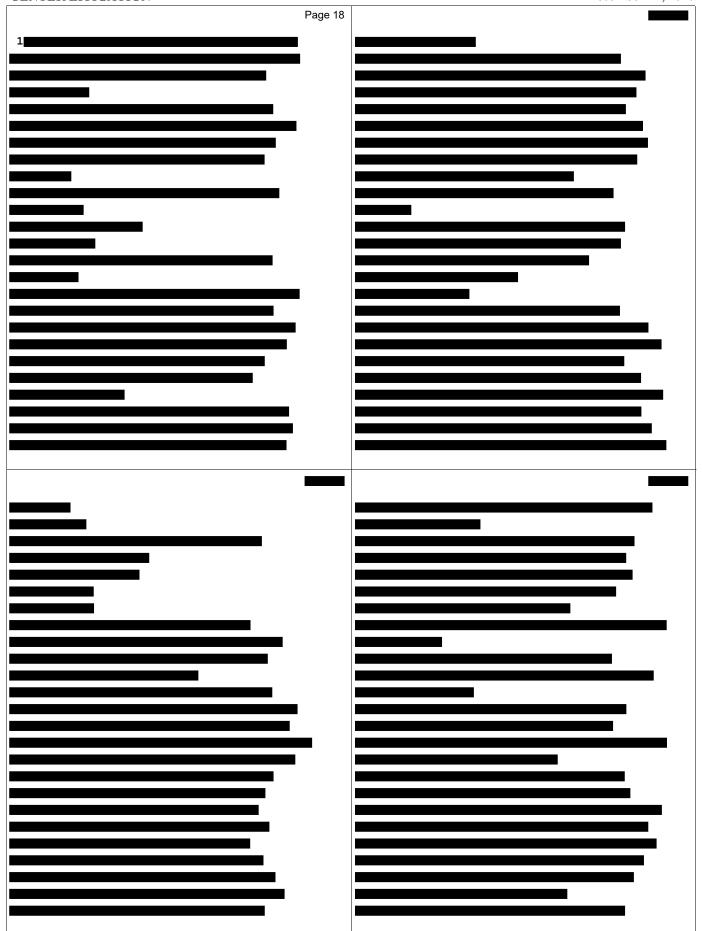
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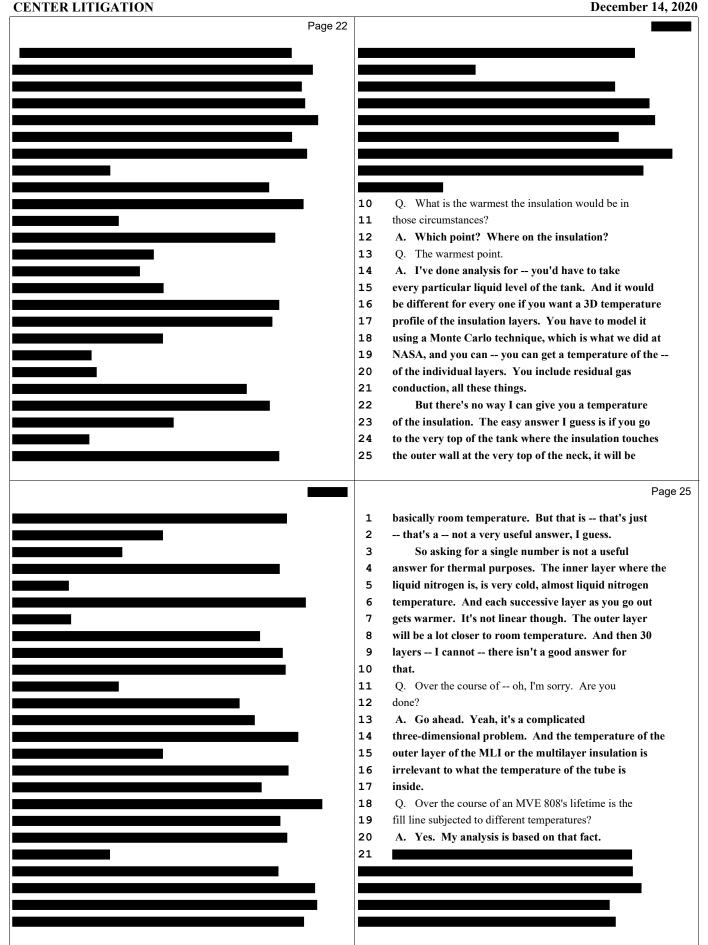


- 2 Q. Both your upper bound and your lower bound are
- 3 estimates; correct?
- 4 A. I don't know that I would call them an
- 5 estimate. How can I put this? An estimate would often
- 6 have an error or a possible error that goes above and 7
 - below.
- 8 In this case what I've done is the actual
- 9 temperature profile in the tube falls somewhere in
- 10 between these two extremes. And so what I've done is
- 11 set an upper limit and a lower limit on the possible
- 12 length change. So it's -- it's they're limiting --
- 13 they're limits. They're upper and lower limits. The
- 14 actual contraction falls between these two limits.
- 15 O. But you've established some assumptions and
- 16 then calculated out what you considered to be the upper
- 17 and lower bounds. And those are not the actual values
- 18 for Tank 4 in operation; correct?
- 19 A. That is not correct. The tank, Tank 4 -- now,
- 20 I have to assume that Tank 4 was operating in a
- 21 condition where it had liquid nitrogen inside the tank,
- 22 and then I have to assume that the -- it's in a room
- 23 that's near room temperature. So there are assumptions
- 24 from that perspective that, you know, if the tank was
- 25 filled with something besides liquid nitrogen, then no,

- 1 a very uncomfortable temperature for workers.
- 2 Q. Are your upper and lower bounds based on actual
- 3 temperature measurements of Tank 4 in use?
- 4 A. Obviously not because I was -- nor was anyone
- 5 -- the temperature of a room can change by a few degrees
- 6 over the course of a day, over the course of weeks. But
- 7 the temperature of the room would be some number that's
- 8 within the range -- within a small range of the number
- 9 I've used.



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negligibly small changes.

Q. And so those small changes in the liquid level

3 would be also making small changes to the thermal

4 profile of the fill line; correct?

A. Incredibly small. Because the -- any

6 significant change would come from a fill event. The

7 liquid level only changes appreciably during a fill

event because you can't create or take away liquid

9 quickly. So liquid boils off slowly. That's part of

the cycle; right? Liquid level goes down. That's part

of the cycle. And then you replace it. And that's a 12 cycle. That's one full cycle. One full thermal cycle.

> So, you know, any small changes are going to be incredibly tiny compared to -- the fill event is the event that's the game changer here because the temperature profile changes in the tube when it's sitting there are incredibly small compared to the temperature profile change when you shock this thing with liquid nitrogen.

You bring in a fill of liquid nitrogen, the entire length of the tube goes from being this temperature profile. It shrinks by 23,000ths. And that's what goes into the analysis because that's the event that brings about the thermal length change we're talking about, not little tiny changes in the liquid

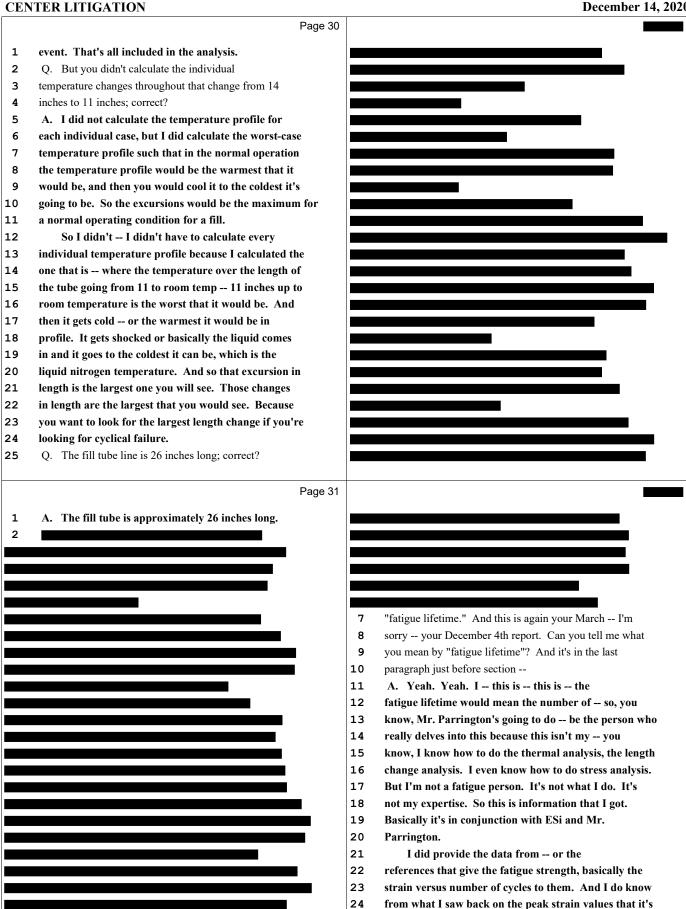
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- 2 Q. And in a tank in normal usage in an IVF lab,
- 3 the portions of the fill line with and without liquid
- 4 nitrogen present versus with gaseous nitrogen present
- 5 would change constantly; correct?
- 6 A. Can you restate the question?
- 7 Q. In an MVE 808 that's in use in an IVF lab, so
- 8 being used to store samples with liquid nitrogen present
- 9 and being replenished regularly, would the portion of
- 10 the fill line with and without liquid nitrogen present
- 11 be in constant flux?
- 12 A. Okay. I think I understand what you're saying.
- 13 So if samples were inserted and samples taken out, there
- 14 would be some small change in the liquid level. Okay?
- 15 But not very much. And then when the tank is filled --
- 16 so if it was at 11 inches and then it was filled to 14
- 17 inches, yes, the level in that tube would change. And
- 18 so it would -- a longer portion of the tube would then
- 19 be cold.
- 20 However, there's only -- it doesn't
- 21 significantly change the assumptions. If the level is
- 22 now at 14 inches, less of the tube is subjected to
- 23 thermal change. So yeah. So there would be small
- 24 changes in the liquid level. I'm not saying there's
- 25 not. But with respect to what I've done here, those are

- level.
- 2 Because there's not enough of the tube involved 3
 - to cause major length change like there is in these fill
- 4 events. The fill events are 23,000ths in length change.
- 5 Q. Liquid nitrogen is constantly evaporating out
- 6 of an MVE 808 that's in use; correct?
- 7 A. Yes.
- 8 Q. And the rate of evaporation may vary, but it's
- 9 always burning off at some level; correct?
- 10 A. Yes, it is.
- 11 Q. When the lid is removed and the tank contents
- 12 are disturbed, the liquid nitrogen will evaporate
- 13 faster; correct?
- 14 A. Yes. The evaporation rate is dependent upon if
- 15 the lid's on or off.
- 16 Q. And based on your testimony within the last few
- 17 minutes, it sounds like you did not calculate for all of
- 18 the fluctuations for the entire life span of Tank 4's
- 19 fill line; correct?
- 20 A. The change -- so the fill event, the thing
- 21 that's a fill event. You consider a cycle for a fill
- 22 event -- includes liquid nitrogen. So if it's at 14
- 23 inches, level goes down to 11 inches, and then there's a
- 24 fill event. Okay? Then let's say some number of days
- 25 later liquid level goes down again, there's a fill

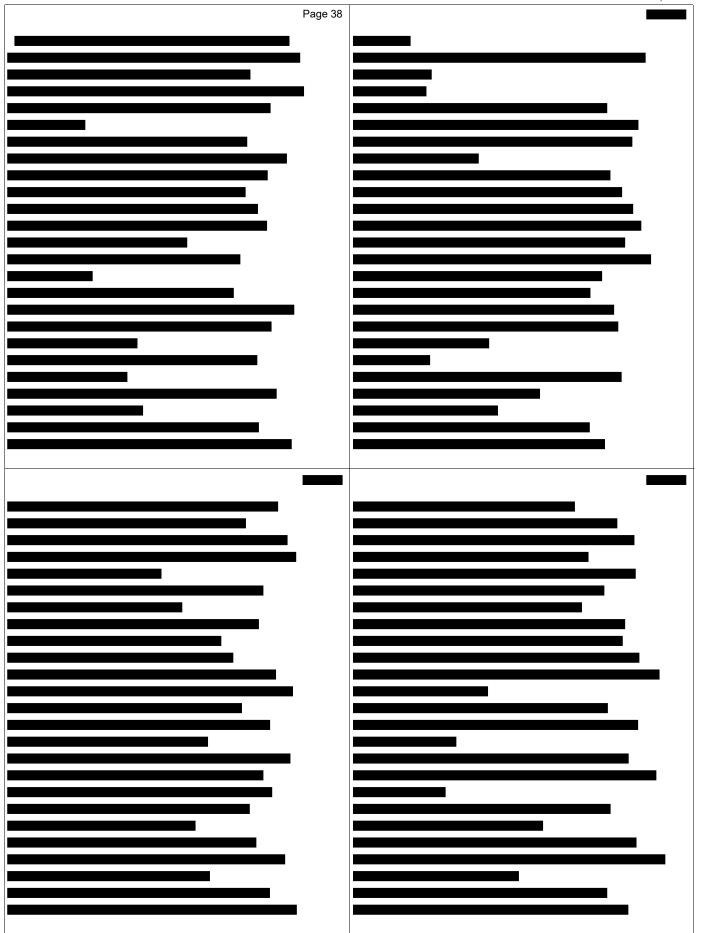


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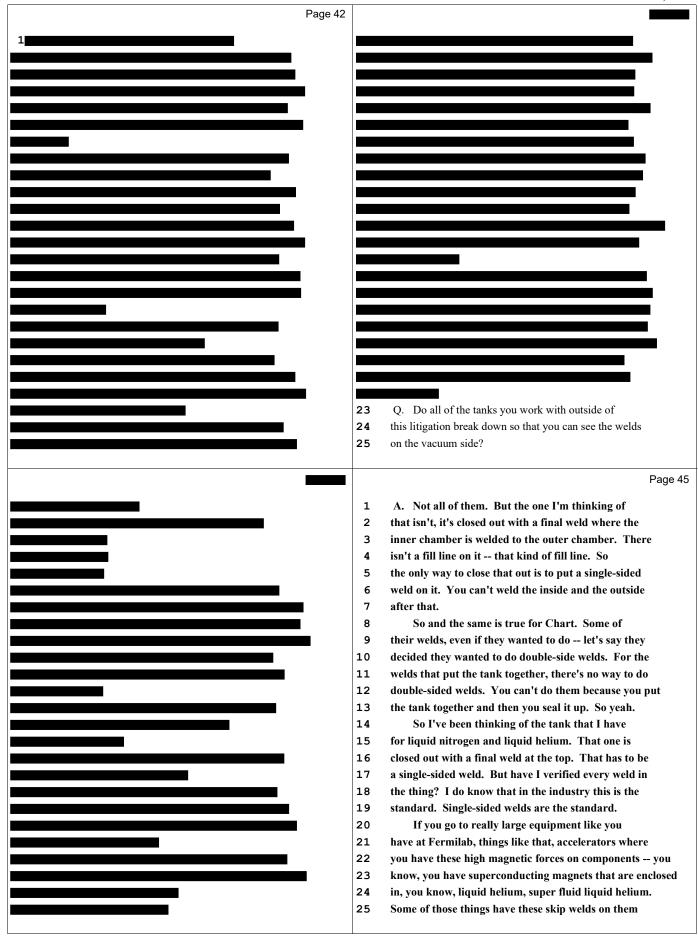
well below the number that allows you to have at least

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Page 34 Page 36 1 10,000 cycles. So it's well past that. It's less than through a calculation of your own? 1 2 2 a third of that value. A. Yes. Just by multiplying the number of days 3 And so that's what I'm saying is the results 3 times at most one cycle per day. So it's an 4 that come from Parrington and ESi say that this is --4 approximation. Go back through the data and show that 5 5 the cycle time would be -- the cycle life at these it's less than that. 6 extreme strains, meaning going from the initial 6 Q. So you think that Tank 4 actually had less than 7 7 condition of the profiling the tank and considering that 2200 fills? 8 8 the liquid level's at 11 inches and considering that you A. It may have. 9 go all the way to nitrogen temperature, that is the 9 Q. Did you analyze the controller data to see how 10 cyclical load, the thermal cyclical load. That's the 10 many times a fill was initiated on Tank 4? 11 11 cyclical load that happens with this tank. And if you A. I didn't count them up. 12 12 Q. Did you analyze the controller data in any way even have that, you know, cyclical load, the tank that 13 -- the failure, the fatigue lifetime is past the 10,000 13 to see how many times a fill was initiated on Tank 4? 14 that the graph goes to. It's beyond that. 14 A. I didn't specifically analyze it for number of 15 15 But the details of that analysis are going to fills. 16 16 be Ron Parrington's to deal with because I didn't do the Q. When you receive a cryogenic tank you purchase 17 fatigue analysis. I'm just reporting based on this 17 from a vendor, do you examine the welds? 18 result from his analysis. 18 A. No. No. I actually usually don't. I do a 19 Q. So this last paragraph before Section B on page 19 leak test on it however. 20 3, does that paragraph contain your opinions or Mr. 20 Q. Is that every single tank you obtain? 21 Parrington's? 21 A. Yes. I'm working in a research environment. 22 22 A. It contains my opinions relevant to what this So yes, we do a leak test on it. I've done it. 23 23 means as far as the number of cycles. But it is based O. So every time you obtain a new cryogenic tank, 24 on the results of the outcome of Parrington's analysis. 24 you do a leak test on it before using it? 25 25 I did not do the analysis. A. Let me -- let me say yes for the research Page 35 Page 37 1 Q. Do you intend to testify before a jury that the 1 dewars. For service dewars, no. No. If we have a --2 fatigue lifetime of the fill port fitting to inner wall 2 you know, a liquid nitrogen dewar that we use for, you 3 weld is in excess of 10,000 cycles? 3 know, transporting liquid from one place to another, no, 4 4 A. I intend to testify that I passed the results we don't check those. They're kind of off-the-shelf 5 5 of my thermal analysis to Ron Parrington, and the items. But for the dewars that I use in-house, ves, I 6 results he gave back indicate that this is true because 6 check them. The ones for experiments. 7 7 that's the actual truth of how this went. I did the MR. DUFFY: We've been going for about an hour, 8 8 thermal analysis. I knew what the boundary conditions Amy. Do you want to just take a five-minute break? 9 were. I passed that off to the thermal analysts -- I 9 MS. ZEMAN: That sounds good. 10 mean, excuse me, the stress -- the FEA people. And they 10 MR. DUFFY: Thank you. 11 did the results and got the results back. So that's the 11 THE VIDEOGRAPHER: We are now going off the 12 extent of what I'll say about that. 12 record at 10:33 a.m. Pacific Standard Time. 13 13 Q. Do you intend to testify before a jury that (Whereupon a break was taken from 10:33 to 14 during the six years that Tank 4 was in service it was 14 15 subjected to approximately 2200 fill line thermal 15 THE VIDEOGRAPHER: Okay. We are now going back 16 16 on the record. And the time is 10:45 a.m. Pacific 17 Standard Time. A. That is an approximate number. And yes, I'll 17 18 -- yes. It's approximate. 19 Q. What's your basis for that opinion? 20 A. That would be if it's filled once every other 21 day. Actually, that's once a day. Yeah, that's once 22 per day. And I don't think it actually was filled once 23 per day if you go back and look at the logs. It's not 24 filled once per day. 25 Q. Did you determine 2200 approximate fills



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CENTER LITIGATION December 14, 2020 Page 46 Page 48 1 because they're buttressing because they have very, very it's in that text or not because that's a cryogenic heat 1 2 2 high magnetic forces on them. transfer text, not a general cryogenic engineering text. 3 But in the kind of tanks we're talking about, I 3 Q. Did you ask if anyone at Chart has tested a 4 4 two-sided vacuum weld? haven't seen two-sided welds. 5 5 Q. Aside from any analysis done by Dr. Kasbekar in A. No, I didn't specifically ask them that 6 this litigation, are you aware of any testing having 6 question. 7 7 been done on the two-sided vacuum welds? Q. Have you reviewed any presale design acceptance 8 8 A. I am aware of the skip welds used in the -- in testing and validation of the single-sided weld used on 9 the Fermilab experiments. I don't know of particular 9 the MVE 808? 10 testing, though, of double-sided welds. 10 A. No. 11 11 Q. Have you searched for any such testing? Q. Are you aware of any such testing and 12 A. Well, I did -- I did do some searching because 12 validation having been completed? A. No. But there's a long history of field use 13 I knew about the -- their -- you know, the large 13 14 equipment has -- you know, how do you weld on support 14 with these welds. So I think that this design 15 15 things with welds on each side? So I knew about that. happened quite a while ago, and there's a long history 16 16 But I didn't run across any in my -- in my of these tanks being in the field with successful use 17 search, but I don't know 100 percent that there's not --17 with a single-side weld. 18 you know, there could be somebody analyzed this and did 18 Q. Do you recall at your prior deposition 19 tests or whatever. I don't know. 19 referring to a tank manufactured by Janis that you were 20 20 But I do know in cryogenics classes when or are currently using or about to use? 21 learning this stuff in grad school one of the things 21 A. Yes. This was the tank that I said -- I think 22 22 we're taught is don't use double-sided welds. There's I said -- I thought it was Janis. I don't know 100 23 23 percent for sure. But I think it's a Janis tank. explanation why -- how to design weld preps or for these 24 seal welds as is -- you know, an example is the -- is 24 Q. Is that tank capable of being dismantled? 25 25 the MVE fitting that's machined down so that the two A. No. That's the one I just talked about that Page 47 Page 49 1 thicknesses are nearly similar. 1 has the closeout weld on the top. 2 So when I was in graduate school I designed all 2 Q. And it does not have a fill line? 3 single-sided welds with weld preps, went to a lot of 3 A. It does not have a -- it does not have a fill 4 4 trouble to machine those pieces in a way that there line that penetrates into the inside space. It has some 5 5 would be a similar thickness between them is part of tubes that penetrate through -- there's a quick cooldown 6 what I was taught and did. 6 coil on it, but it doesn't have a tube that goes into 7 7 Q. Are you saying you were specifically instructed the inner space, no. It has a tube penetration through 8 8 to not use double-sided welds on vacuum vessels? the top that goes and there's a coil that wraps around 9 A. Yes. By my Ph.D. advisor and my co-advisor. 9 the tank, the inner tank, that you can use for --10 Yes. 10 O. Does that tube connect to the inner vessel? 11 O. Were those verbal instructions? 11 Does that tube connect to the inner vessel? 12 A. Yes. And this is what's taught in, you know, 12 A. Not in the sense that it brings liquid into the 13 13 cryogenics class. Yeah. inner vessel. It's -- there's a tube wrapped around the Q. Specifically people are told do not use 14 14 inner vessel that you can use if you're going to cool

15 double-sided welds on vacuum --

16 A. Because of the potential of virtual leaks.

17 Q. Do you have any literature about that?

18 A. It is probably in some of the cryogenics texts

19 that I have.

20 Q. Is it in any of the literature that you relied

21 on for your reports?

22 A. It may be in the cryogenics texts that I relied

23 on for the latest report, the second rebuttal report.

24 But I didn't cite that particular -- I'm not sure if

25 it's in that text or not. I'll have -- I'm not sure

15 down to liquid helium temperature, you can run liquid

16 nitrogen through that first to cool it down. But it

17 doesn't have a manual fill line as you would call it.

18 It's an annular heat exchanger is what it is.

19 Q. But at some point does it attach to the inner

20 vessel?

21 A. Physically, yes. Not on a fluid connection.

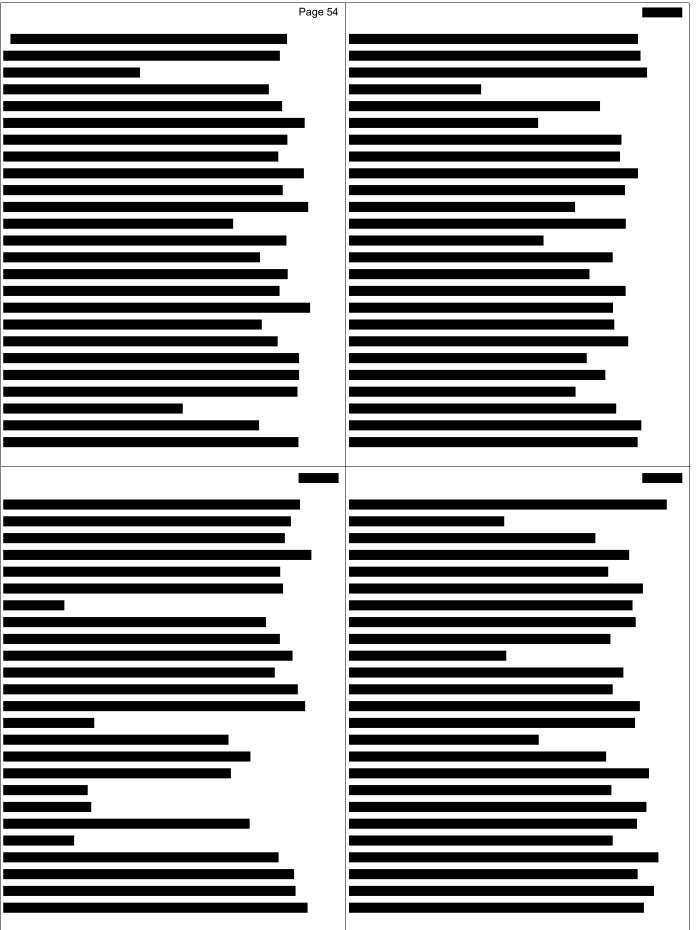
22 Q. It doesn't have an opening into the inner

23 vessel: correct?

24 A. Yes.

25 Q. Could water pass through a breach between two

Page 50 Page 52 1 cold spaces if one space was at a slightly higher it will flash and become vapor. But relieve and the 1 2 2 pressure than the other space? other word, that doesn't -- that doesn't work. Those 3 A. I'm not sure what you - how water relates to 3 don't work. 4 this. Water would be in solid form in this particular 4 Q. But to reduce the pressure is an accurate 5 5 concept? 6 Q. That wasn't my question to this specific case 6 A. No, not to reduce the pressure. It's because 7 7 or to Tank 4 to liquid nitrogen. The question stands as of the reduced pressure. Yeah. 8 8 Q. Okay. Thanks. In your opinion Tank 4 deformed 9 A. Could water pass through an opening from one 9 because the sieve material released nitrogen it had 10 space to another if there's a pressure difference? Is 10 previously absorbed; correct? 11 11 that the question? A. That is correct. 12 12 Q. Correct. Q. And in other words, the sieve off-gassed A. Yes. That's what drives the flow is pressure 13 13 nitrogen and pressurized the vacuum space? 14 14 A. Yes. As it warmed. 15 15 Q. Could liquid nitrogen pass through a breach Q. As the pressure increased in the vacuum space 16 16 as the sieve off-gassed, would the sieve had reabsorbed between two cold spaces if one space was at a slightly higher pressure than the other? 17 17 some of the nitrogen? 18 A. Of course it could. What's really going to be 18 A. So the -- really wouldn't work quite that way. 19 -- this is a different situation from water though. And 19 There's an equilibrium that occurs between -- as the 20 20 when you ask the water question, it really depends on temperature rises there's an equilibrium that occurs 21 what the temperature is and what the pressure is. 21 which involves pressure, temperature to the pressure over the getter or the vapor pressure over the top of 22 If you're getting at could liquid nitrogen pass 22 23 23 through and go into the vacuum space, it could. But it the getter, and the temperature related to one another. 24 would be a vapor when it got to the other side to the 24 As that getter gets to 300 -- or, you know, up 25 25 lower pressure. near room temperature, there's very little gas that can Page 51 Page 53 1 Q. Did you say that it would be a vapor when it 1 remain inside the getter. There's a little bit, but got to the other side to relieve the pressure? 2 2 very little nitrogen gas will stay inside the getter at 3 A. No. That's not what I said. 3 that point. So I don't know that it really reabsorbs. 4 4 Q. Okay. My question is what did you say then? It's just there would be a little bit of gas that would 5 5 I'm just trying to make sure I understood you correctly. not come out of the getter, a minimal amount. Very few 6 A. The liquid would go -- when it -- in the 6 of those little binding sites inside the liquid 7 7 process of going from the high pressure to the low nitrogen -- excuse me -- inside the getter would retain 8 8 pressure, it would vaporize. nitrogen atoms -- or molecules, actually. Molecules. 9 Q. Let me try again. I -- really, I honestly want 9 So yeah, it's not really that it reabsorbs it. It just 10 to know what the word was because it wasn't picked up in 10 reaches an equilibrium. An equilibrium. 11 the transcript. So I think you said something like, but 12 it would be a vapor when it got to the other side to 13 something the pressure. What's the word that you used? 14 A. I'm not sure. 15 O. It might have been relieve? Does that sound 16 right? 17 A. No, not relieve the pressure. 18 O. Or to release? 19 A. No. 20 Q. Okay. 21 A. Because no, neither of those two words make 22 sense. Because of the reduced pressure? I don't know. 23 Q. Okay. 24 A. That actually makes sense. Because of the 25 reduced pressure. Because there's reduced pressure so



CENTER LITIGATION December 14, 2020 Page 58 Page 60 1 So I -- it really depends. materials that you've had firsthand experience 1 2 2 Q. What is the extremely valuable material that overseeing this cryogenic storage of. 3 you have firsthand experience overseeing the storage of? 3 A. I think those are the most valuable ones. And 4 A. Yeah, once again extremely valuable depends on 4 by storage, I mean, it's not storage in the sense that 5 5 your -- but I had materials in dewars. And I don't know it's sitting there actively doing nothing. But it's, 6 that it was storage, but it was testing at low 6 you know, cold temperature for weeks and we need to 7 7 maintain it. So it's more than storage. It's not just, temperature. And a failure of that dewar could have 8 8 destroyed the hardware that eventually flew on an x-ray oh, we're dropping this in a tank and leaving it there. 9 astrophysics mission. By the way, the mission -- that 9 It's we're doing testing at the same time. So it's more 10 spacecraft failure was on Japanese spacecraft that had 10 than just storage. 11 11 another kind of failure. But so they lost it anyway. Q. Do you want to amend the sentence in your 12 12 But that insert, those -- that cooler was worth -- to report to cover more than storage? 13 13 rebuild it would have been multiple millions -- multiple A. I don't know if storage and testing is relevant 14 million dollars. I don't know exactly how much. But 14 to this case. I mean, you know, because there's not 15 15 that kind of hardware, I consider that very valuable. cryogenic testing in this particular case. 16 16 Maybe that's not very valuable to some people, but to me Q. So you're saying only storage is relevant? 17 that was very valuable. 17 A. To this particular case. 18 Q. Was it the dewar that was valuable to you or 18 Q. Okay. So is there any other extremely valuable 19 the materials in that dewar? 19 material that you've had firsthand experience in 20 A. The materials. The, you know, 20 overseeing the cryogenic storage of? 21 one-of-a-kind-in-the-world cooler for space flight 21 A. Let me think. The most valuable ones are the 22 mission. It cooled an x-ray detector for --22 James Webb and the x-ray mission hardware. 23 23 O. So what were the materials that were extremely But as I said before, there are valuable things 24 valuable in that dewar? 24 here at the University. They're not at the same level. 25 25 A. Superconducting magnets, paramagnetic You know, we're talking tens of thousands of dollars, Page 59 Page 61 1 1 materials. Just the construction in the development of not millions of dollars. So it's not the same. 2 that was, you know, years, decades in development at 2 Q. Do you have any firsthand experience overseeing 3 NASA. 3 the cryogenic storage of human tissue? 4 Q. Is there any other extremely valuable material 4 A. I think I answered this in the last deposition. 5 5 that you have firsthand experience in overseeing the No, I don't.

6 cryogenic storage of? 7 A. Well, I wouldn't -- I wouldn't -- once again, 8 I'm not saying cryogenic storage. But I'm saying 9 cryogenic -- you know, so it was cooled down and tested. 10 If there was some failure in the middle of the test, it 11 could have destroyed it. There were components for the 12 James Webb space telescope testing. 13 Q. To clarify, though, you are saying -- I'm 14 reading from your report at the top of page 5 where you 15 write "I have firsthand experience in overseeing the 16 cryogenic storage of extremely valuable material...," 17 etcetera.

So my question is I'm trying to find out from

you what extremely valuable material is it that you have

storage of? So so far you've identified superconducting

had firsthand experience in overseeing the cryogenic

identified components for the James Webb telescope

magnets and another related materials. You just

Are there any other extremely valuable

6 Q. Do you think embryologists have experience 7 working with cryogenic storage of human tissue? 8 A. Yes. Hope so. 9 Q. Do you think embryologists have experience 10 working with extremely valuable materials? 11 A. Yes. 12 Q. Do you consider yourself an ordinary user of 13 cryogenic tanks like the MVE 808? 14 A. That's a tough question because I'm sitting 15 here as a cryogenics expert. So of course I'm not an 16 ordinary user. 17 Q. Have you ever operated a cryogenic tank without 18 a controller? 19 A. Yes. In cases where I was just doing some 20 short-term testing and the material would tolerate 21 warming up. For the cases where we were talking about a 22 while ago where it's more expensive material, there is a

THE REPORTER: Amy, can you pause for just one

minute? I don't need to go off the record, I just need

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controller.

December 14, 2020 Page 62 Page 64 1 to plug something in real quick. Thank you. Thank you. They know how important it is. But it's something they 1 2 2 (Discussion off the record.) use, not something they're expert in. So that's my 3 MR. DUFFY: We're a little over an hour now. 3 definition of ordinary user. 4 THE REPORTER: Sure. Just two minutes? 4 Q. Is your definition of "their ordinary users" 5 5 MR. DUFFY: Yeah. everyone who utilizes cryogenic tanks? 6 **THE VIDEOGRAPHER:** Off the record at 11:35 a.m. 6 A. I don't know if it's everyone who uses -- in 7 7 this particular context, I don't know if it's everybody Pacific Standard Time. 8 8 (Whereupon a break was taken from 11:35 to who uses cryogenic tanks. Some people use cryogenic 9 11:46.) 9 tanks to, you know, carry liquid nitrogen from one place 10 THE VIDEOGRAPHER: We are now going back on the 10 to another and fill a cold trap or cool down a device 11 11 record. The time is 11:46 a.m. Pacific Standard Time. that they use. They may have a cold station on an SEM 12 Q. BY MS. ZEMAN: Dr. Miller, how would systematic 12 or something like that. There they use cryogenic 13 checks and procedures ensure continuous and 13 fluids, you know, for a particular use; but they're not 14 uninterrupted function of a cryogenic tank? 14 keeping some material at a cryogenic temperature, you 15 15 A. Could you repeat the question? I'm sorry. know, for a long time. 16 16 Q. Sure. How would systematic checks and So I'm thinking of the person who's utilizing 17 procedures ensure continuous and uninterrupted function 17 cryogenics -- not necessarily a cryogenic expert, but 18 of a cryogenic tank? 18 utilizing cryogenic environment for their testing, for 19 A. If you maintain the regular checks and regular 19 their storage, whatever it is. And they understand 20 - of the tank, you'll know what's going on with it. 20 cryogenic tanks and their use and what they're used for 21 And if there are any problems, you can get them -- they 21 and how they -- how they're important to the function of 22 22 could be dealt with and serviced. their particular task. 23 23 O. So they won't necessarily prevent a tank from So I don't just mean anybody who would pick up 24 malfunctioning, but they would allow the end user to 24 a tank of liquid nitrogen and carry it over to the cold 25 25 address the malfunction; is that correct? trap, you know, and pour it in for their pumping system. Page 63 Page 65 1 1 That's a little different from someone who's tasked A. Yes. The tank could be -- you know, if there was an alarm that wasn't working properly, if it's 2 2 with, you know, doing their work in a cryogenic 3 regularly checked it could be corrected and prevent, you 3 environment and have some material. 4 4 know -- prevent there not being, you know, a control or Q. How do you know that ordinary users view 5 5 alarm or systems working. cryogenic tanks as requiring systematic checks and 6 Q. If you could look at the last sentence on page 6 procedures to ensure continuous and uninterrupted 7 7 4 of your December 4th report. And that sentence then function? 8 8 runs onto and ends at the top of page 5. And once A. Like I said, I've worked in an environment 9 you've looked at that sentence, could you tell me who 9 where there were users of cryogenics, people who 10 you mean by the term "their ordinary users"? 10 characterize materials, did optics tests to cryogenic 11 A. People who work with cryogenic systems. You 11 temperatures, but were not cryogenic engineers. They 12 know, in the -- in the world I worked in, not everybody 12 were ordinary users of cryogenic technology tanks, 13 13 at NASA who uses cryogenic systems is a cryogenics vessels to carry out their tasks. But they certainly 14 expert. They were people that I worked with who are 14 understood. 15 15 experts in other kinds of areas, optics experts, That they may not have known how to revac a 16 materials experts. And they used cryogenic systems in 16 system. They may not have known, you know, how to 17 their -- in their work. And so I would call them 17 design welds or how to build a tank or put out a 18 18 ordinary users of cryogenics systems. specification for a tank, but they did understand that 19 19 They're not, you know, people who design they needed to keep their material cold and that they 20 20 needed to check on it regularly and to have monitoring cryogenic systems. They're not people who build them. 21 21 They're not designing cryogenic systems. But it's part for their tanks because they understood the value of 22 22 what was in there. of the work that they do, and they utilize cryogenic 23 tanks every day in their -- in their -- in the work 23 Q. Does measuring the liquid nitrogen in a tank

they're doing. So they would be ordinary users because

they're not -- they're involved in using cryogenics.

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daily constitute a systematic check?

A. It is systematic in that if it's implemented

A. I don't know how Chart would be able to

require. As I said before, there's no governing body

that requires this. I don't know how they would require

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A. I think it had a slow leak over possibly

several years, which is not unusual for a tank. And the

getter managed that leak in a way that allowed the tank

December 14, 2020 Page 70 Page 72 1 to continue to perform its function, maintain insulation particular material. They're synthetic zeolites. 1 2 2 integrity. And only when the tank liquid nitrogen level And this paper Yang and Burris is a synthetic 3 went to zero did the vacuum level inside the tank rise. 3 zeolite. It is by a different manufacturer, but there's 4 4 And the implosion happened a day and a half later. not -- data at cryogenic temperatures is not done at 300 5 5 So yeah, the tank was capable of and did psi for this material. It's -- there is -- there is 6 maintain the safety of the tissue in there as long as it 6 data in the paper for 300 psi, but it's not for the 7 7 was filled. But if the tank is not filled, it cannot cryogenic temperatures. I mean --8 8 Q. Wasn't that study done specifically to obtain maintain cryogenic temperatures. 9 MS. ZEMAN: I would like to enter your rebuttal 9 data covering higher pressures and temperatures? 10 report dated December 11th as Plaintiffs' Exhibit 441. 10 A. Absolutely. Because a lot of -- some uses of 11 11 Philip, could you circulate that, please. this material are for pressure swing adsorption, like if 12 THE VIDEOGRAPHER: Absolutely. Just bear with 12 you want to separate one species from another. There's 13 13 me here. some cases where you can use this for nitrogen oxygen 14 (Plaintiffs' Exhibit 441 marked for 14 separation. When.... 15 15 identification.) THE REPORTER: You're cutting -- you're cutting 16 16 THE VIDEOGRAPHER: Okay. It's now loaded in out just a little bit. There's some cases where you can 17 17 use this --18 Q. BY MS. ZEMAN: Dr. Miller, if you could take a 18 THE WITNESS: Okay. I'll slow down. 19 look at that exhibit and then let me know if you 19 MR. DUFFY: No. I think it's actually just 20 20 recognize it. your connection. 21 21 A. Yeah, this is my report dated December 11th. THE REPORTER: It's your -- it's your 22 Q. In the second paragraph on page 5 of your 22 connection 23 23 December 11 report you refer to a paper. Is that paper MR. DUFFY: It's just starting to get a little 24 the Yang and Burris paper titled "Nitrogen adsorption 24 under water. 25 25 isotherms for zeolite and activated carbon"? THE REPORTER: Right. Right. Page 71 Page 73 1 THE WITNESS: I'm on an Ethernet, but okay. 1 A. Second paragraph. Which page? 2 Q. Page 5. 2 So these materials are also used to do pressure 3 A. Oh, okay. Second paragraph of the report. 3 swing adsorption which is where you raise the pressure 4 4 and one species preferentially adsorbs on the material. Okav. Yes. 5 5 Q. That is the Yang and Burris paper that you're You then pump out, and then you lower the pressure. And 6 referring to in that paragraph? 6 you basically can separate one species from another that 7 7 A. Yes. way. So one gets adsorbed, you can pump out the other 8 8 Q. What do you mean when you write that "...the species, and then you get a concentration of that 9 paper included data for material at high pressure but 9 species. 10 the data from the paper used for" your "analysis was 10 So yes, they were doing measurements for those 11 from low temperature and" low "pressure experiments..."? 11 applications, but they also made measurements at low 12 A. Yeah. So they -- there's data in the paper for 12 temperature. And I think one of the reasons is they 13 13 78.9 kelvin approximately. That data is not at 300 psi. want to look at what the absolute capacity is at low 14 If you -- there's a nondimensional graph in there. If 14 temperature for each of these binding sites. You know, 15 15 vou take the data and vou take it back to dimensional how many -- how many molecules can you get into this 16 form, the data is not from 300 psi. You don't do 16 material basically. How many N2 molecules can you get 17 adsorption data for nitrogen on zeolites. 17 into the material in a particular set of conditions. 18 18 So yes, at a cursory glance the paper is for And by the way, Siliporite is a brand name for 19 19 a zeolite made by, comma, trying to remember the name, 300 psi. But when you dig into the paper, there's data 20 that is -- the reason in the last deposition I didn't 20 there for low temperature. And it's -- it was useful, I 21 21 know what Siliporite is, in the cryogenics world it's thought, because it had a -- it was -- there's not a lot 22 22 marketed as Cryo-Sieve. And then in the gas and oil of data out there on these type of synthetic zeolites at 23 industry it's marketed under the brand name registered 23 low temperature because they are often used for other 24 24

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trade name Siliporite. So there's a -- they're the same

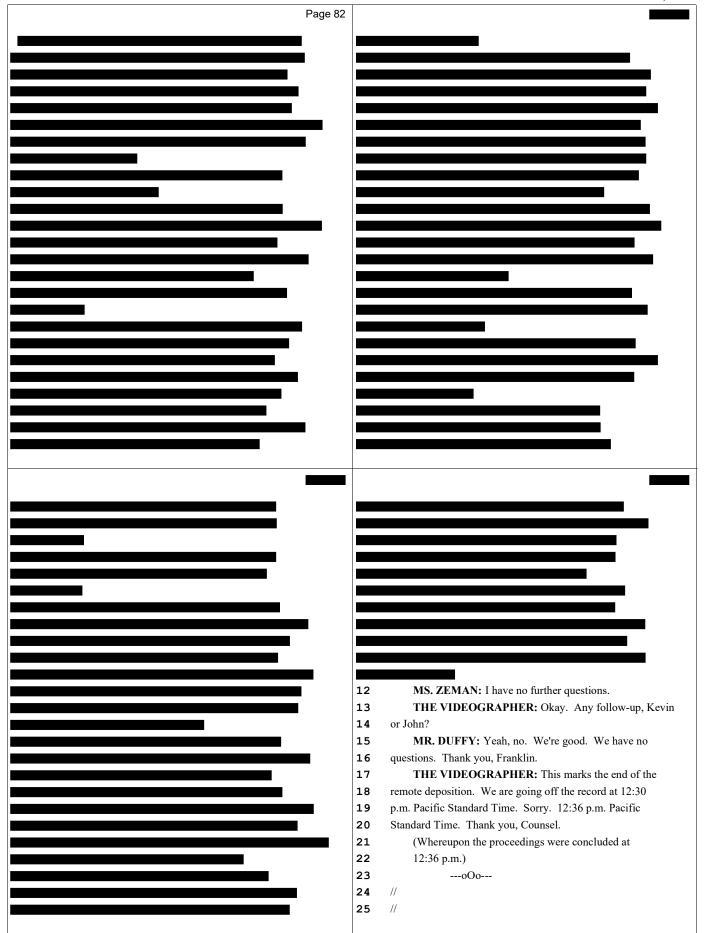
material. They're a class of materials. It's not a

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Now, there's a difference of opinion here. Dr.

	TER LITIGATION		December 14, 2020
	Page 74		Page 76
1	Kasbekar thinks you can't use one brand for another, you	1	divided by T. And so because you know the temperature
2	can't swap them out. I think it would be 10 percent.	2	is 78.92, you can back out what the pressure is. And
3	I maybe even better than that. But that's okay. I	3	all these points are at low pressure. There's nothing
4	made a measurement to confirm that the tank the	4	at high pressure for the 78.92.
5	getter will, in fact, hold as much gas as I thought it	5	Q. Okay.
6	would.	6	A. That calculate what the pressure is. It's low
7	So yeah, if you read the paper carefully, if	7	pressure. It's not it's not at high pressure.
8	you go in and take had the data that's on the graph,	8	Q. And is that data from Figure 5 the only data
9	take it from nondimensional form and put it back into	9	you used from this paper?
10	dimensional form, you'll see that it is at low pressure.	10	A. Yes. I think so.
11	Q. BY MS. ZEMAN: There is some data in the paper	11	Q. Does this paper discuss desorption?
12	at low pressure. But it's more than a cursory glance;	12	A. Does it discuss what?
13	correct? I mean, the paper was done because high	13	Q. Desorption.
14	pressure adsorption data for a broad temperature range	14	A. Desorption. No, I don't think so. It's
15	are required to perform a quantitative system analysis;	15	they're adsorption isotherms is what. Why
16	correct?	16	Q. Is it your opinion that the Cryosiev material
17	A. Yes. But as a part of that analysis, they	17	in the MVE 808 tanks releases 100 percent of the
18	did they did work at low temperatures.	18	molecules it's absorbed at room temperature?
19	Q. Sure.	19	A. It depends on the pressure that's in the tank.
20	A. So, you know, I was looking through papers for	20	It's probably about 5 percent that would stay in the
21	data at this temperature. And this it was in there.	21	in the sieve possibly, about 5 percent. That won't make
22	You know, some other papers had no low temperature data,	22	any difference in whether there's enough gas in there to
23 24	low pressure data. Q. Where in the Yang and Burris paper is the data	23 24	collapse the tank. Q. And does this paper explain whether sieve would
25	for low temperature and pressure experiments?	25	release 95 percent of the molecules at room temperature?
23	for fow temperature and pressure experiments:	25	release 75 percent of the molecules at foolin temperature.
	Page 75		Page 77
1	Page 75 A. There's a graph.	1	Page 77 A. It explains it, but This particular paper
1 2	-	1 2	
	A. There's a graph.		A. It explains it, but This particular paper
2	A. There's a graph. MS. ZEMAN: Philip, could you enter the Yang	2	A. It explains it, but This particular paper does not have room temperature data. But it does have
2	 A. There's a graph. MS. ZEMAN: Philip, could you enter the Yang and Burris paper as Exhibit 442. 	2	A. It explains it, but This particular paper does not have room temperature data. But it does have some 273 K data. And what you can see is that at one
2 3 4	A. There's a graph. MS. ZEMAN: Philip, could you enter the Yang and Burris paper as Exhibit 442. THE VIDEOGRAPHER: Yes. (Plaintiffs' Exhibit 442 marked for identification.)	2 3 4 5 6	A. It explains it, but This particular paper does not have room temperature data. But it does have some 273 K data. And what you can see is that at one atmosphere the delta M over M for this material, it's .0064 grams per gram. Whereas the material would hold, like, point one — so it's in the third decimal place
2 3 4 5	A. There's a graph. MS. ZEMAN: Philip, could you enter the Yang and Burris paper as Exhibit 442. THE VIDEOGRAPHER: Yes. (Plaintiffs' Exhibit 442 marked for identification.) THE VIDEOGRAPHER: Okay.	2 3 4 5 6 7	A. It explains it, but This particular paper does not have room temperature data. But it does have some 273 K data. And what you can see is that at one atmosphere the delta M over M for this material, it's .0064 grams per gram. Whereas the material would hold, like, point one so it's in the third decimal place the amount it will hold compared to what it will hold
2 3 4 5 6 7 8	A. There's a graph. MS. ZEMAN: Philip, could you enter the Yang and Burris paper as Exhibit 442. THE VIDEOGRAPHER: Yes. (Plaintiffs' Exhibit 442 marked for identification.) THE VIDEOGRAPHER: Okay. Q. BY MS. ZEMAN: Dr. Miller, if you could open up	2 3 4 5 6 7 8	A. It explains it, but This particular paper does not have room temperature data. But it does have some 273 K data. And what you can see is that at one atmosphere the delta M over M for this material, it's .0064 grams per gram. Whereas the material would hold, like, point one so it's in the third decimal place the amount it will hold compared to what it will hold at cryogenic temperature. And this is at 273, not at
2 3 4 5 6 7 8 9	A. There's a graph. MS. ZEMAN: Philip, could you enter the Yang and Burris paper as Exhibit 442. THE VIDEOGRAPHER: Yes. (Plaintiffs' Exhibit 442 marked for identification.) THE VIDEOGRAPHER: Okay. Q. BY MS. ZEMAN: Dr. Miller, if you could open up Exhibit 442 and let me know if that is the Yang and	2 3 4 5 6 7 8	A. It explains it, but This particular paper does not have room temperature data. But it does have some 273 K data. And what you can see is that at one atmosphere the delta M over M for this material, it's .0064 grams per gram. Whereas the material would hold, like, point one so it's in the third decimal place the amount it will hold compared to what it will hold at cryogenic temperature. And this is at 273, not at 293. So it will be even less than that. So it's like 6
2 3 4 5 6 7 8 9	A. There's a graph. MS. ZEMAN: Philip, could you enter the Yang and Burris paper as Exhibit 442. THE VIDEOGRAPHER: Yes. (Plaintiffs' Exhibit 442 marked for identification.) THE VIDEOGRAPHER: Okay. Q. BY MS. ZEMAN: Dr. Miller, if you could open up Exhibit 442 and let me know if that is the Yang and Burris paper that you are referring to?	2 3 4 5 6 7 8 9	A. It explains it, but This particular paper does not have room temperature data. But it does have some 273 K data. And what you can see is that at one atmosphere the delta M over M for this material, it's .0064 grams per gram. Whereas the material would hold, like, point one so it's in the third decimal place the amount it will hold compared to what it will hold at cryogenic temperature. And this is at 273, not at 293. So it will be even less than that. So it's like 6 percent. But if you go up to 290 kelvin, it's going to
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	Page 78	
-	A Committed on Table with Albertain About	
1	A. Sure it does. It's right here in the data	
2	right there. Only but not you've seen my rebuttal	
3	report. It doesn't depend much on temperature at 77	
4	kelvin at nitrogen temperature. But at a higher	
5	temperatures, yes, it does.	
6	Q. Is Cryosiev identical to molecular sieve type	
7	13X?	
8	A. I don't know that it's identical. It's	
9	similar. They're different manufacturers. One's Union	
10	Carbide the other is, you know, Arkema. All of these	
11	things fall under the classification of zeolites. But	
12	there could be small differences.	
13	Q. And could have different dimensions; correct?	
14	A. They can have different dimensions, yes.	
15	Q. Could they have different granular density?	
16	A. They could. That's why I did the test. Just	
17	to get rid of any	
18	Q. Do they have different micropore sizes?	
19	MR. DUFFY: Do you want to finish the answer?	
20	I'm sorry. Objection.	
21	THE WITNESS: So that's why I did the test	
22	because the two brands are going to be different and may	
23	be different in some cases. They could be. You know,	
24	they're these materials are kind of bound into	
25	pellets. I don't think there's a significant as in	_
	Page 79	
	Page 79	
1	one's going to absorb twice as much as another one.	
1 2	one's going to absorb twice as much as another one. It's not going to happen like that. There could be some	
	one's going to absorb twice as much as another one. It's not going to happen like that. There could be some small under 10 percent difference in performance maybe.	
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	Page 86					Page 88
1	I have read the foregoing deposition	1	of the testin	mony given	by the witness. (Fed. R. Civ. P.
2	transcript and by signing hereafter, subject to	2	30(f)(1)).			
3	any changes I have made, approve same.	3	Bef	ore comple	tion of the deposi	tion, review of
4		4	the transcri	ipt [XX] w	as [] was not :	requested. If
5	Dated .	5		_	ges made by the	_
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